



CHEM11042 *Fundamentals of Chemistry*

Term 1 - 2018

Profile information current as at 15/09/2025 08:33 am

All details in this unit profile for CHEM11042 have been officially approved by CQUniversity and represent a learning partnership between the University and you (our student). The information will not be changed unless absolutely necessary and any change will be clearly indicated by an approved correction included in the profile.

General Information

Overview

This unit will provide you with the fundamental principles of chemistry that underpin the medical sciences and provide a strong foundation on which you can develop an understanding of biochemistry and molecular science. You will gain an appreciation of the nature of matter, classic atomic structure and how energy is involved in bond formation. These concepts will be developed to explain the forces between molecules that govern chemical interaction. You will be introduced to the chemistry of electrolytes, acids, bases and buffers. This study will be supported by simple calculations to assist you in relating to the pH scale. The study of organic chemistry and molecules central to the life sciences will enable you to develop an understanding of the biochemistry and molecular biology relevant to your specific discipline. The naming and classifying of chemical compounds will enable you to be conversant with accepted scientific terms. Tutorials and on-line activities will complement the theoretical knowledge gained in lectures and provide you with the basic mathematical and analytical tools required in the application of chemistry to your specific discipline.

Details

Career Level: *Undergraduate*

Unit Level: *Level 1*

Credit Points: 6

Student Contribution Band: 8

Fraction of Full-Time Student Load: 0.125

Pre-requisites or Co-requisites

There are no requisites for this unit.

Important note: Students enrolled in a subsequent unit who failed their pre-requisite unit, should drop the subsequent unit before the census date or within 10 working days of Fail grade notification. Students who do not drop the unit in this timeframe cannot later drop the unit without academic and financial liability. See details in the [Assessment Policy and Procedure \(Higher Education Coursework\)](#).

Offerings For Term 1 - 2018

- Distance
- Rockhampton

Attendance Requirements

All on-campus students are expected to attend scheduled classes – in some units, these classes are identified as a mandatory (pass/fail) component and attendance is compulsory. International students, on a student visa, must maintain a full time study load and meet both attendance and academic progress requirements in each study period (satisfactory attendance for International students is defined as maintaining at least an 80% attendance record).

Website

[This unit has a website, within the Moodle system, which is available two weeks before the start of term. It is important that you visit your Moodle site throughout the term. Please visit Moodle for more information.](#)

Class and Assessment Overview

Recommended Student Time Commitment

Each 6-credit Undergraduate unit at CQUniversity requires an overall time commitment of an average of 12.5 hours of study per week, making a total of 150 hours for the unit.

Class Timetable

[Regional Campuses](#)

Bundaberg, Cairns, Emerald, Gladstone, Mackay, Rockhampton, Townsville

[Metropolitan Campuses](#)

Adelaide, Brisbane, Melbourne, Perth, Sydney

Assessment Overview

1. **Online Quiz(zes)**

Weighting: 30%

2. **Written Assessment**

Weighting: 20%

3. **Examination**

Weighting: 50%

Assessment Grading

This is a graded unit: your overall grade will be calculated from the marks or grades for each assessment task, based on the relative weightings shown in the table above. You must obtain an overall mark for the unit of at least 50%, or an overall grade of 'pass' in order to pass the unit. If any 'pass/fail' tasks are shown in the table above they must also be completed successfully ('pass' grade). You must also meet any minimum mark requirements specified for a particular assessment task, as detailed in the 'assessment task' section (note that in some instances, the minimum mark for a task may be greater than 50%). Consult the [University's Grades and Results Policy](#) for more details of interim results and final grades.

CQUniversity Policies

All University policies are available on the [CQUniversity Policy site](#).

You may wish to view these policies:

- Grades and Results Policy
- Assessment Policy and Procedure (Higher Education Coursework)
- Review of Grade Procedure
- Student Academic Integrity Policy and Procedure
- Monitoring Academic Progress (MAP) Policy and Procedure – Domestic Students
- Monitoring Academic Progress (MAP) Policy and Procedure – International Students
- Student Refund and Credit Balance Policy and Procedure
- Student Feedback – Compliments and Complaints Policy and Procedure
- Information and Communications Technology Acceptable Use Policy and Procedure

This list is not an exhaustive list of all University policies. The full list of University policies are available on the [CQUniversity Policy site](#).

Previous Student Feedback

Feedback, Recommendations and Responses

Every unit is reviewed for enhancement each year. At the most recent review, the following staff and student feedback items were identified and recommendations were made.

Feedback from Moodle forums and student- staff discussions

Feedback

The twice daily response to the Q&A forum was appreciated by students

Recommendation

Continue good practice with twice daily sessions to answer any questions on the Q&A forum. Students will be encouraged to read other posts on this forum.

Feedback from Moodle forums and student-staff discussions

Feedback

Generally students found the structure of this Unit very workable. The students appreciated the regular format for each week with included: lecture scenario and focus notes for the week, lecture slides and videos incorporating use of the document camera for 'ease of following', reading and tutorial question list along with worked solutions and videos of same. Students appreciated the availability of lectures well ahead of time as this enabled them to plan studies around work commitments.

Recommendation

Continue good practice and ensure frequent use of document camera for all recordings by all teaching staff so as to show live workings for calculations and explanations.

Feedback from Moodle forums and student-staff discussions

Feedback

Students enjoyed the scenario based approach for the introduction of topics for this Unit. This approach was also extended to the Poster assessment item which was well received.

Recommendation

The scenario based approach to this Unit will continue and further scenarios will be introduced.

Feedback from Moodle forums and student-staff discussions

Feedback

More face to face tutorials for all students

Recommendation

The face to face tutorials for the Rockhampton campus will remain. In addition, an investigation into the possibility of a non-compulsory workshop for Flex students to attend Rockhampton campus will be undertaken. The possibility of frequent zoom sessions will also be considered, although large cohort numbers may make these session difficult.

Feedback from Moodle forums

Feedback

Some students did not like the prescribed online text,

Recommendation

Students were recommended to purchase a preferred option of the online text however this was not always taken. A hard copy text was also available. Future offerings should highlight to students that the hard copy is available.

Feedback from Moodle forums and student-staff discussions

Feedback

Some students made comment that there was too much content in this Unit or the Maths was too difficult.

Recommendation

Students will be encouraged to contact the ALC early in the term if they feel that their basic maths skills are not of appropriate standards. Required skills are multiplication, division, and basic algebra including rearranging equations. The unit content can be broken down to 3 broad topics and this will be highlighted to students early on in the term. The topics introduced have been reduced both in quantity and depth from previous offerings with a focus now on relevance and application to paramedic/chiropractic scenarios.

Unit Learning Outcomes

On successful completion of this unit, you will be able to:

1. Apply concepts of atomic structure to explain molecular bonding and nuclear reactivity.
2. Apply chemical concepts to healthcare situations.
3. Identify categories of organic compounds and their potential chemical interactions.
4. Perform basic chemical calculations.

Alignment of Learning Outcomes, Assessment and Graduate Attributes

 N/A Level	 Introductory Level	 Intermediate Level	 Graduate Level	 Professional Level	 Advanced Level
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Alignment of Assessment Tasks to Learning Outcomes

Assessment Tasks	Learning Outcomes			
	1	2	3	4
1 - Online Quiz(zes) - 30%	•		•	•
2 - Written Assessment - 20%		•		
3 - Examination - 50%	•		•	•

Alignment of Graduate Attributes to Learning Outcomes

Graduate Attributes	Learning Outcomes			
	1	2	3	4
1 - Communication	•	•	•	•
2 - Problem Solving	•	•	•	•
3 - Critical Thinking	•	•	•	•
4 - Information Literacy	•	•	•	•
5 - Team Work				
6 - Information Technology Competence				
7 - Cross Cultural Competence				
8 - Ethical practice				
9 - Social Innovation				
10 - Aboriginal and Torres Strait Islander Cultures				

Alignment of Assessment Tasks to Graduate Attributes

Assessment Tasks	Graduate Attributes									
	1	2	3	4	5	6	7	8	9	10
1 - Online Quiz(zes) - 30%		•	•	•		•				
2 - Written Assessment - 20%	•	•	•	•		•				

Assessment Tasks	Graduate Attributes									
	1	2	3	4	5	6	7	8	9	10
3 - Examination - 50%	•	•	•	•						

Textbooks and Resources

Textbooks

CHEM11042

Prescribed

Chemistry - Core Concepts

Edition: 1st (2016)

Authors: Allan Blackman, Adam Bridgeman, Gwendolyn Lawrie, Daniel Southam, Christopher Thompson, Natalie Williamson

Wiley

Brisbane , Queensland , Australia

ISBN: 9780730311065 (paperback); 9780730324935 (Etext)

Binding: Other

Additional Textbook Information

You have the option to purchase an E-Text: If you chose to purchase the E-Text please ensure you select the option that includes the Wiley Plus Learning space as it contains important tutorial content.

Blackman et al, Chemistry Core Concepts, 1st edition, 2016 with Wiley Plus Learning Space.

This item can be purchased by accessing the link on the CHEM11042 Moodle site.

The E-Text provides animations and short lecture style videos to assist you in your learning and the Wiley Plus Learning Space provides additional quizzes to further reinforce the course material and to assist you to gauge your understanding of the concepts presented.

In addition, students are required to have a Scientific calculator (preferred brand SHARP EL-531XH) and Periodic table (optional)

[View textbooks at the CQUniversity Bookshop](#)

IT Resources

You will need access to the following IT resources:

- CQUniversity Student Email
- Internet
- Unit Website (Moodle)

Referencing Style

All submissions for this unit must use the referencing styles below:

- [American Psychological Association 6th Edition \(APA 6th edition\)](#)
- [Vancouver](#)

For further information, see the Assessment Tasks.

Teaching Contacts

Aoife Power Unit Coordinator

a.power@cqu.edu.au

Schedule

Week 1 - 05 Mar 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: How this unit will run L2: Introduction to matter L3: Measurement	1.1 Why study chemistry? 1.2 Introduction to matter 3.3 Measurement	All weekly quizzes must be completed by the Wednesday of the following week, i.e. week 1's quiz should be completed by the Wednesday of week 2 Weekly on-line quizzes Due: Orientation Week Monday (5 Mar 2018) 11:45 pm AEST

Week 2 - 12 Mar 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Atomic structure L2: Periodic Table and ion formation L3: Scientific notation and the metric system	1.4 The structure of the atom 1.5 The Periodic Table 3.1 Section: Expression of large and small numbers 3.1 Section: Exponents and logarithms 3.2 Section: SI Units	Quiz 1 closes 11.55pm, Wednesday 14th March 2018

Week 3 - 19 Mar 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Molecular formula and the mole L2: Lewis dot diagrams L3: Ionic bonding	2.1 Representations in chemistry 4.4 The mole 6.1 Fundamentals of bonding 6.2 Ionic bonding	Quiz 2 closes 11.55pm, Wednesday 21st March 2018

Week 4 - 26 Mar 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Covalent Bonding and Lewis dot diagrams L2: VSEPR L3: Intermolecular forces	6.3 Lewis structures 6.4 VSEPR 7.2 Intermolecular forces	Quiz 3 closes 11.55pm, Wednesday 28th March 2018

Week 5 - 02 Apr 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Solutions and concentration L2: Molarity and stoichiometry L3: Dilutions	10.1 Solutions and solubility 10.2 Section: Concentration of solutions 10.2 Section: Diluting a solution	Quiz 4 closes 11.55pm, Wednesday 4th April 2018

Vacation Week - 09 Apr 2018

Module/Topic	Chapter	Events and Submissions/Topic
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Week 6 - 16 Apr 2018

Module/Topic	Chapter	Events and Submissions/Topic
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L1: Chemical reactions
L2: Energy in reactions
L3: Equilibrium

4.1 Chemical and physical change
4.2 Chemical equations
4.3 Balancing chemical equations
4.6 Stoichiometry, limiting reagents and percentage yield
8.3 Enthalpy
13.4 Temperature dependence of chemical reactions
13.5 Reaction mechanism and catalysis
9.1 Chemical equilibria
9.2 The equilibrium constant, and the reaction quotient
9.4 How systems at equilibrium respond to change

Quiz 5 closes 11.55pm, Wednesday 18th April 2018

Week 7 - 23 Apr 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Acid bases and neutralisation reactions L2: pH calculations L3: Buffers	11.1 The Bronsted-Lowry definition 11.5 The molecular basis of acid strength 11.2 Acid-base reactions in water 11.6 Buffer solutions	Quiz 6 closes 11.55pm, Wednesday 25th April 2018

Week 8 - 30 Apr 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: The Ideal Gas equation L2: Dalton's Law and Henry's Law L3: Nuclear chemistry	7.3 Gases 7.4 Gas mixtures Supplementary material	Quiz 7 closes 11.55pm, Wednesday 2nd May 2018 Poster: Applications of chemistry to Paramedic/Chiropractic scenarios. Due: Week 8 Friday (4th May 2017) 11:45 pm AEST Poster: Applications of Chemistry to Paramedic or Chiropractic Scenarios Due: Week 8 Friday (4 May 2018) 11:45 pm AEST

Week 9 - 07 May 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Introduction to organic chemistry and alkanes L2: Alkenes and alkynes L3: Aromatic compounds	2.1 Representations in chemistry 2.2 Section: Naming organic compounds - alkanes 14.1 Introduction to hydrocarbons 14.2 Alkanes 14.4 Reactions of alkanes 14.3 Alkenes and alkynes 14.5 Reactions of alkenes 14.6 Aromatic compounds	Quiz 8 closes 11.55pm, Wednesday 9th May 2018

Week 10 - 14 May 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Functional groups, aldehydes, ketones and ethers L2: Alcohols L3: Oxidation of alcohols	2.2 Section: Naming organic compounds - functional groups 15.2 Alcohols 15.3 Reactions of alcohols	Quiz 9 closes 11.55pm, Wednesday 16th May 2018

Week 11 - 21 May 2018

Module/Topic	Chapter	Events and Submissions/Topic
L1: Amines L2: Carboxylic acids and derivatives L3: Chiral compounds	15.4 Amines 15.6 Carboxylic acids 15.7 Nomenclature of carboxylic acids and derivatives 15.1 Chiral compounds	Quiz 10 closes 11.55pm, Wednesday 23rd May 2018

Week 12 - 28 May 2018

Module/Topic	Chapter	Events and Submissions/Topic
Review		

Review/Exam Week - 04 Jun 2018

Module/Topic	Chapter	Events and Submissions/Topic
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Exam Week - 11 Jun 2018

Module/Topic	Chapter	Events and Submissions/Topic
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Assessment Tasks

1 Weekly on-line quizzes

Assessment Type

Online Quiz(zes)

Task Description

This assessment is comprised of ten on-line quizzes (for Weeks 1 to 10). Completing these quizzes will give you an indication of your understanding of the concepts presented each week and encourage you to stay on track with your study. This assessment requires you to apply the concepts to answer a series of multiple choice questions. Each quiz is comprised of 10 multiple-choice questions selected from a question bank. All questions in each quiz are of equal value and each of the 10 quizzes will contribute 3%, totaling 30% for this assessment item. The quiz is not timed and you are allowed three attempts; the highest score of the three attempts will be recorded. Note that questions are generated randomly and you will receive different questions on subsequent attempts. There is a minimum mark requirement of 40% for this assessment item, i.e., you must attain 40% in total for the 10 quizzes in order to pass the unit overall.

Number of Quizzes

10

Frequency of Quizzes

Weekly

Assessment Due Date

Orientation Week Monday (5 Mar 2018) 11:45 pm AEST

Each quiz will close at 11:55pm on the Wednesday of the following week. i.e. The week 1 quiz will close on the Wednesday of week 2.

Return Date to Students

Quiz results will be released after the completion of each attempt. Answers will be released after the quiz has closed.

Weighting

30%

Minimum mark or grade

40% of total marks available for the 10 quizzes

Assessment Criteria

All questions are of equal weighting. One mark will be awarded for each correct response. Incorrect responses will not incur a penalty.

Referencing Style

- [American Psychological Association 6th Edition \(APA 6th edition\)](#)
- [Vancouver](#)

Submission

Online

Submission Instructions

Complete each week's quiz by following the link on the Moodle site.

Learning Outcomes Assessed

- Apply concepts of atomic structure to explain molecular bonding and nuclear reactivity.
- Identify categories of organic compounds and their potential chemical interactions.
- Perform basic chemical calculations.

Graduate Attributes

- Problem Solving
- Critical Thinking
- Information Literacy
- Information Technology Competence

2 Poster: Applications of Chemistry to Paramedic or Chiropractic Scenarios

Assessment Type

Written Assessment

Task Description

This assessment requires you to design a professional poster that clearly illustrates the importance of chemistry in a scenario or medical condition that is significant to healthcare, particularly in your chosen field of paramedic or chiropractic science. The poster should be well presented and include an image or diagram that represents and highlights the significance of the scenario chosen.

You are to use your chemical knowledge and conduct research to explain the scenario and also suggest ways of improving, treating or preventing the issue being discussed.

The poster should be designed using PowerPoint application. A template for the poster and a 'how-to' video will be made available on the Moodle site. Further task details and a list of potential topics will also be available on the units Moodle site.

Assessment Due Date

Week 8 Friday (4 May 2018) 11:45 pm AEST

Return Date to Students

Week 11 Friday (25 May 2018)

Weighting

20%

Minimum mark or grade

40%

Assessment Criteria

A marking Rubric will be provided on the Moodle site for this unit. Marks will be awarded for:

1. Introduction to the topic
2. The use of an appropriate image
3. The chemical explanation of the topic
4. Discussion on the potential of chemical knowledge to help explain/solve the problem presented
5. Overall presentation
6. Grammar and spelling
7. Referencing

Referencing Style

- [American Psychological Association 6th Edition \(APA 6th edition\)](#)
- [Vancouver](#)

Submission

Online

Submission Instructions

Submit your poster as a single .pdf file

Learning Outcomes Assessed

- Apply chemical concepts to healthcare situations.

Graduate Attributes

- Communication
- Problem Solving
- Critical Thinking

- Information Literacy
- Information Technology Competence

Examination

Outline

Complete an invigilated examination.

Date

During the examination period at a CQUniversity examination centre.

Weighting

50%

Length

120 minutes

Minimum mark or grade

40%

Exam Conditions

Closed Book.

Materials

Dictionary - non-electronic, concise, direct translation only (dictionary must not contain any notes or comments).

Calculator - all non-communicable calculators, including scientific, programmable and graphics calculators are authorised

Academic Integrity Statement

As a CQUniversity student you are expected to act honestly in all aspects of your academic work.

Any assessable work undertaken or submitted for review or assessment must be your own work. Assessable work is any type of work you do to meet the assessment requirements in the unit, including draft work submitted for review and feedback and final work to be assessed.

When you use the ideas, words or data of others in your assessment, you must thoroughly and clearly acknowledge the source of this information by using the correct referencing style for your unit. Using others' work without proper acknowledgement may be considered a form of intellectual dishonesty.

Participating honestly, respectfully, responsibly, and fairly in your university study ensures the CQUniversity qualification you earn will be valued as a true indication of your individual academic achievement and will continue to receive the respect and recognition it deserves.

As a student, you are responsible for reading and following CQUniversity's policies, including the [Student Academic Integrity Policy and Procedure](#). This policy sets out CQUniversity's expectations of you to act with integrity, examples of academic integrity breaches to avoid, the processes used to address alleged breaches of academic integrity, and potential penalties.

What is a breach of academic integrity?

A breach of academic integrity includes but is not limited to plagiarism, self-plagiarism, collusion, cheating, contract cheating, and academic misconduct. The Student Academic Integrity Policy and Procedure defines what these terms mean and gives examples.

Why is academic integrity important?

A breach of academic integrity may result in one or more penalties, including suspension or even expulsion from the University. It can also have negative implications for student visas and future enrolment at CQUniversity or elsewhere. Students who engage in contract cheating also risk being blackmailed by contract cheating services.

Where can I get assistance?

For academic advice and guidance, the [Academic Learning Centre \(ALC\)](#) can support you in becoming confident in completing assessments with integrity and of high standard.

What can you do to act with integrity?



Be Honest

If your assessment task is done by someone else, it would be dishonest of you to claim it as your own



Seek Help

If you are not sure about how to cite or reference in essays, reports etc, then seek help from your lecturer, the library or the Academic Learning Centre (ALC)



Produce Original Work

Originality comes from your ability to read widely, think critically, and apply your gained knowledge to address a question or problem